

# Bench-Scale Development of a Non-Aqueous Solvent (NAS) CO<sub>2</sub> Capture Process for Coal-Fired Power Plants (DE-FE0013865)

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# Non-Aqueous Solvent (NAS) Development Pathway

	Previous Work			DOE ARPA-E Project		DOE NETL Project (Current)		Future Development		
Yr	2009-10			2010-13		2014-15		2016-20		2020+
TRL	1	2	3	4		5		6	7	8 & 9

Proof of  
Concept/Feasibility

Pre-Commercial  
Demonstration

## Lab-scale Development (*Previous*)

- Solvent screening to identify promising solvent formulations
- Lab-scale evaluation of NAS Process
- Preliminary technical and economic assessments



## Large Bench-scale System / Relevant Environment Testing (*Current*)

- Finalize NAS formulation
  - Address evaporative losses and solvent costs
- Develop critical process components
  - NAS wash / recovery section
  - NAS regenerator
- Bench-scale testing with in a process unit with major process components
- Demonstrate  $\leq 2,000$  kJ/kg CO<sub>2</sub> using bench-scale system
- Detailed solvent degradation and preliminary emissions studies
- Detailed Techno-Economic & EH&S Assessments
  - Demonstrate T&EA competitiveness and environmental permitability



# R&D Strategic Approach

## Breakdown of the Thermal Regeneration Energy Load

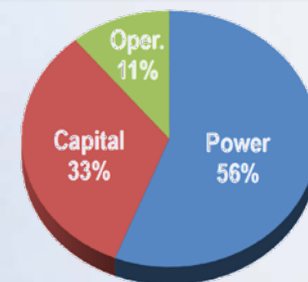
$$q_R = \left[ \frac{C_p(T_R - T_F)}{\Delta\alpha} \cdot \frac{M_{sol}}{M_{CO_2}} \cdot \frac{1}{x_{sol}} \right] + \left[ \Delta H_{v,H_2O} \cdot \frac{p_{H_2O}}{p_{CO_2}} \cdot \frac{1}{M_{CO_2}} \right] + \left[ \frac{\Delta H_{abs,CO_2}}{M_{CO_2}} \right]$$

Reboiler Heat Duty      Sensible Heat      Heat of Vaporization      Heat of Absorption

Solvent	$C_p$ [J/g K]	$\Delta h_{abs}$ [kJ/mol]	$\Delta h_{vap}$ [kJ/mol]	$x_{sol}$ [mol solv./ mol sol'n]	$\Delta\alpha$ [mol CO <sub>2</sub> / mol solv.]	Reboiler Duty [GJ/tonne CO <sub>2</sub> ]
MEA (30%)	3.8	85	40	0.11	0.34	3.22
Lower Energy Solvent System	↓	↓	↓	↑	↑	↓
NAS	1.3	65	1	0.3	0.3	1.71

## Path to Reducing ICOE and Cost of CO<sub>2</sub> Avoided

- Primarily focus on reducing energy consumption – reboiler duty
- Reduce capital expenditure
  - Simplify process arrangement
  - Materials of construction
- Limit operating cost increase



For NAS, heat of vaporization of water becomes a negligible term to the heat duty  
 Process capable of achieving these criteria will have a lower energy penalty than SOTA processes

<sup>1</sup> Rochelle, G. T. Amine Scrubbing for CO<sub>2</sub> Capture. *Science* **2009**, 325, 1652-1654.

# Project Objectives and Technical Challenges

**Objective:** Continue the advancement of the NAS CO<sub>2</sub> Capture Process

- Address specific challenges facing technical and economic potential
- Bench-scale demonstration of the potential to reduce the energy penalty to <2,000 kJ/kg of CO<sub>2</sub> captured

## Specific Challenges

- Minimize solvent losses and make-up
- Solvent degradation and emission studies
- Develop and evaluate process modifications
- Bench-scale evaluation of the NAS CO<sub>2</sub> capture process

**Timeframe:** 10/1/13 to 03/30/15 (BP1, 18 months) 04/1/15 to 06/30/16 (BP2, 15 months)

**Cost:** \$1.51 M BP1, \$1.55 M BP2



RTI NAS Solvent

## Brief Recap of BP1 Achievements

BP1 Achievements	Select Points
Incorporated non-volatile hydrophobic diluent with suitable properties	<ul style="list-style-type: none"> <li>• Vapor pressure &lt;0.13 kPa, 25°C</li> <li>• Low cost</li> <li>• Low viscosity (~2 cP)</li> </ul>
Formulated diluent with hydrophobic amines	<ul style="list-style-type: none"> <li>• Low heats of absorption</li> <li>• No precipitates</li> <li>• Low viscosities (25-30 cP rich)</li> <li>• Reasonable CO<sub>2</sub> capacity</li> <li>• Cost is &lt;\$50/kg</li> </ul>
Demonstrated emissions of NAS below 10 ppm	<ul style="list-style-type: none"> <li>• Designed wash section at lab scale</li> <li>• ~20 ppm emitted without wash section</li> </ul>
Performed long-term evaluation of NAS at lab scale with simulated flue gas containing 13.3% CO <sub>2</sub> , 7.5% H <sub>2</sub> O, 2% O <sub>2</sub> , 50 ppm SO <sub>2</sub> , and balance N <sub>2</sub>	<ul style="list-style-type: none"> <li>• Capture efficiency (~90%)</li> <li>• Long-term, stable operation demonstrated (~100 hrs)</li> </ul>
Completed long-term thermal and oxidative degradation studies at SINTEF	<ul style="list-style-type: none"> <li>• Five week evaluations</li> <li>• Single components of diluent are thermally stable</li> <li>• Carbamate polymerization products not formed</li> <li>• Corrosion results promising (Fe, Ni, Cr)</li> <li>• Eliminated one NAS amine due to severe oxidative degradation</li> </ul>

# BP2 Focus: Bench-scale Testing of Refined Solvents

## Absorber

3" Sch. 10 SS316  
(8.5 m height)  
Mellapak 350X

Temp: 30-55° C  
Pressure: Up to  
200 kPa

Gas Vel: 0.33-1.5  
m/s

L: 15-75 kg/h

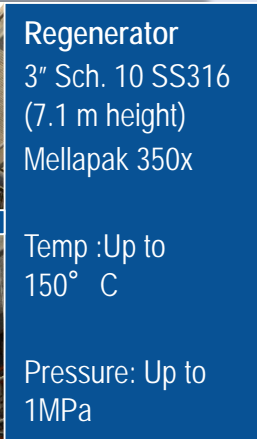


## Regenerator

3" Sch. 10 SS316  
(7.1 m height)  
Mellapak 350x

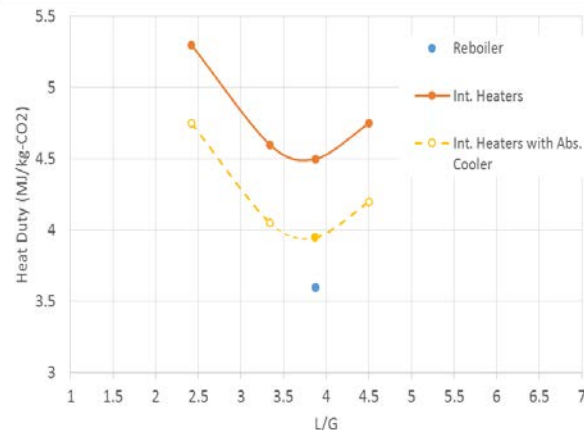
Temp :Up to  
150° C

Pressure: Up to  
1MPa



## Simulated Flue Gas Properties

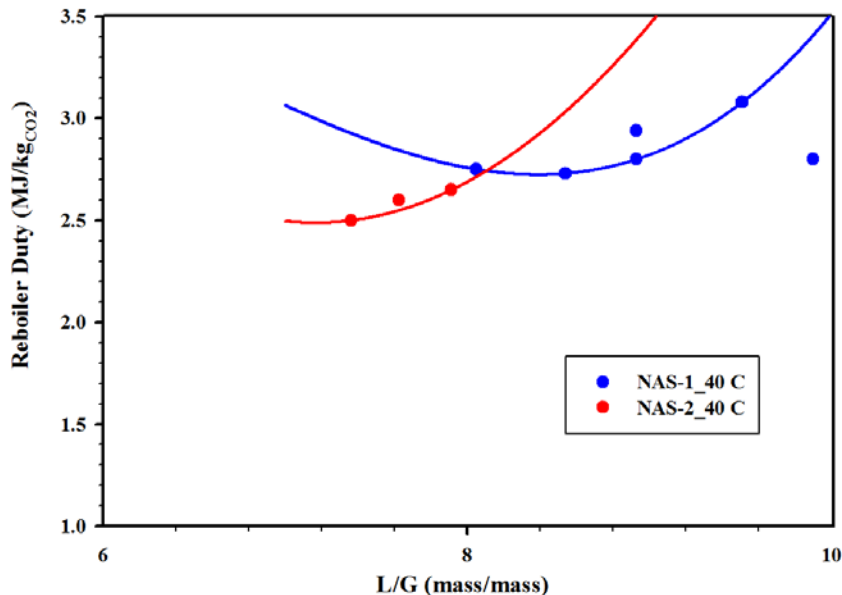
FG Flow Rate:	100 to 485 SLPM
CO <sub>2</sub> Feed Rate:	1.8 to 8.6 kg/h
Feed Temp.:	30 to 50°C
Target Comp:	CO <sub>2</sub> : 13.3%; H <sub>2</sub> O: 6.1%; O <sub>2</sub> : 2.35%; N <sub>2</sub> : bal.
CO <sub>2</sub> Content:	up to 20 %vol
Water Content:	~0 to 12.3%vol



Baseline testing with aqueous MEA

# Bench Scale Test Unit Results with Dry Flue Gas

## REBOILER DUTY (MJ/kg<sub>CO2</sub>) FOR DIFFERENT AMINE CONCENTRATIONS



### Experimental Conditions

Feed Composition:

15.4% CO<sub>2</sub>

Balance N<sub>2</sub>.

100 SLPM

Liquid flows: 0.8-1.9 kg/min

Gas Flows: 0.1-0.21 kg/min

Regenerator Pressure 2.5 bar

90% CO<sub>2</sub> capture efficiency

NAS-1

Absorber: 40° C

Regenerator: 115° C

Interstage Heater Regeneration

NAS-2

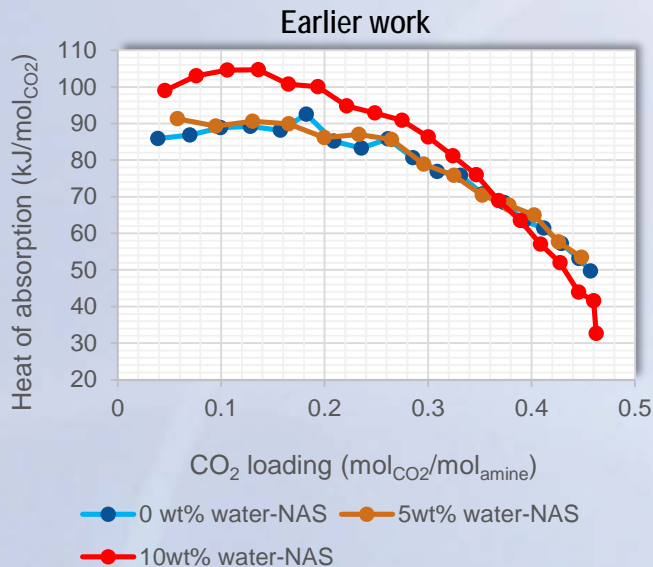
Absorber: 37-40° C

Regenerator: 120-122° C

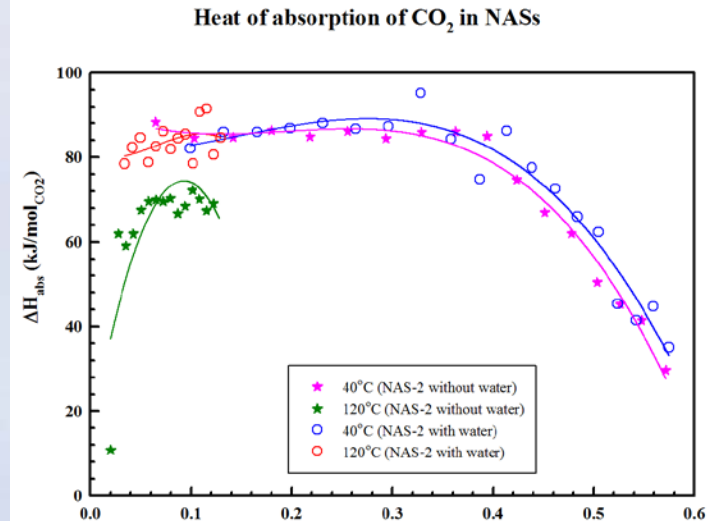
Interstage Heater Regeneration

- Working capacities were lower than anticipated, ~0.15-0.21 moles CO<sub>2</sub>/ mole amine for NAS-1
- Improved slightly for NAS-2 due to slightly lower absorber temperature and higher regenerator temperature
- Still higher than expected based on theoretical values and not a major improvement over other technologies
- Early in our experience at operating the NAS system

# Impact of Water



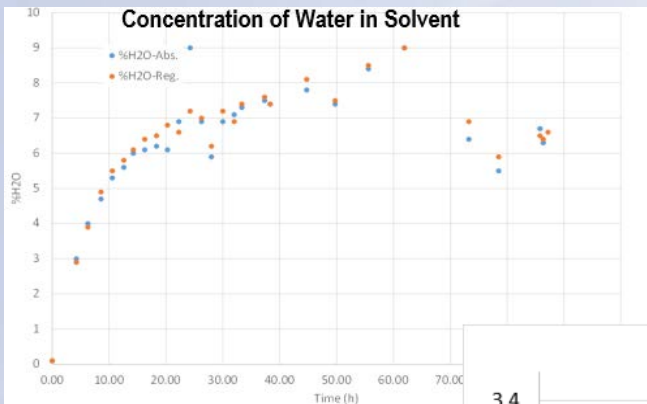
- Observed impact on enthalpy of reaction in earlier NAS formulation
- Measurements at 40° C
- Concentration of water at 10% raised heat of absorption substantially
- Concerns about this impact on reboiler heat duty



- Measured heat of absorption of dry NAS-2 vs. "wet" NAS-2 at 120° C
- Observed increase in heat of absorption when NAS was saturated with water at 120° C
- Expect reboiler duty would go up due to higher  $\Delta h_{abs}$
- Impact on the process is that [water] may need to be kept low. Water becomes separate phase > ~9 wt%
- Increasing the hydrophobicity of the solvent chemistry was thought to be one way to handle



# Bench Scale Test Unit Results with Wet Flue Gas



- 100 hr test with wet flue gas
- Feed composition
  - 15.4% CO<sub>2</sub>
  - 7% H<sub>2</sub>O
  - balance of air
- No water separators
- Water in solvent controlled by absorber temperature profile

## NAS-1

Absorber: 40-36° C

Regenerator: 115 - 95° C

Regenerator Pressure: 2.5-3.75 bar

Interstage Heater Regeneration

## NAS-2

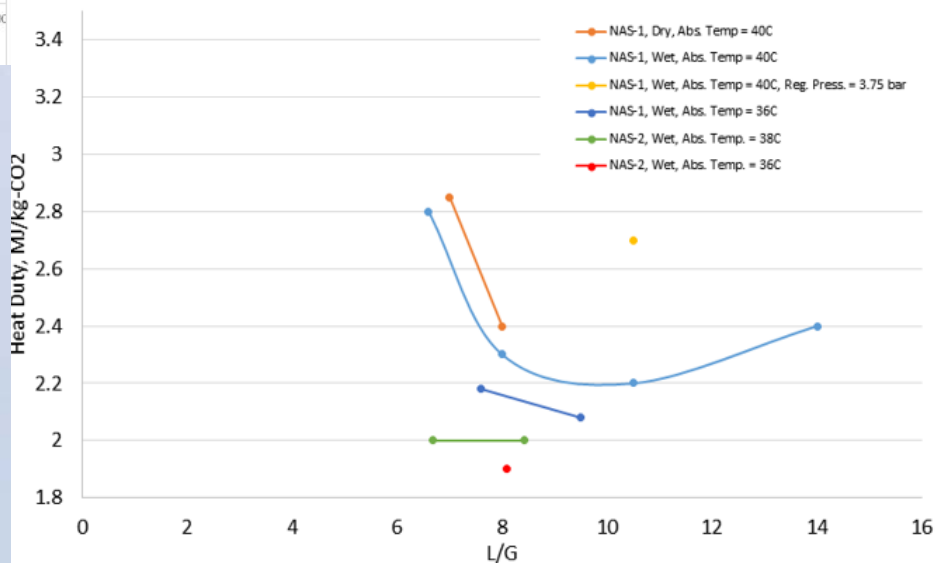
Absorber: 37-40° C

Regenerator: 100-105° C

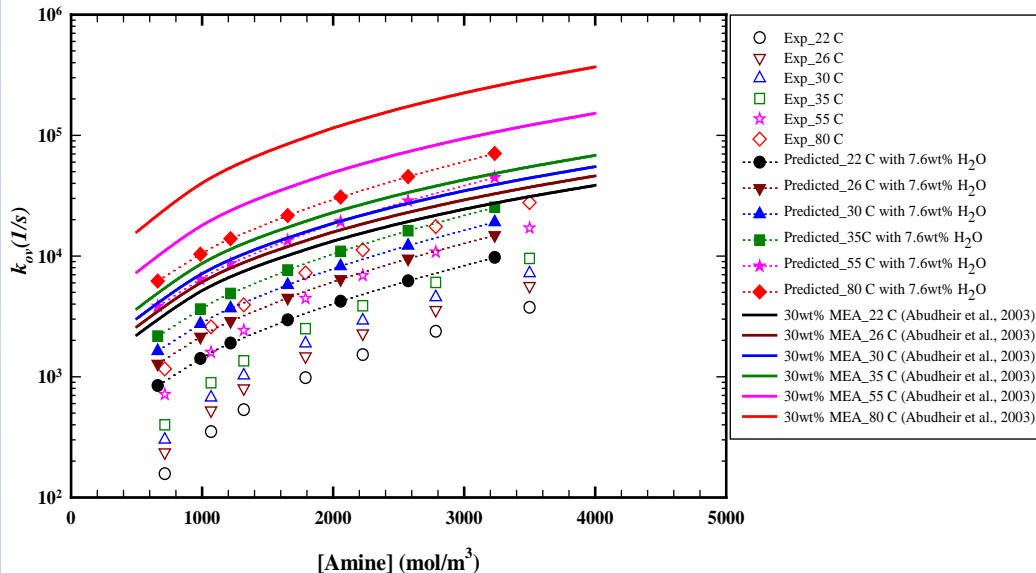
Regenerator Pressure: 2.5 bar

Interstage Heater Regeneration

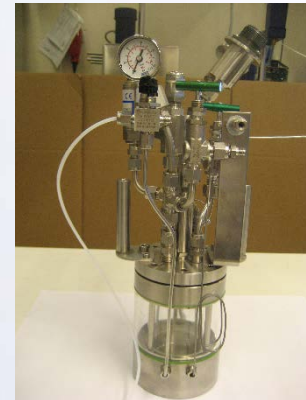
## Regenerator Heat Duty



## Reaction Kinetics

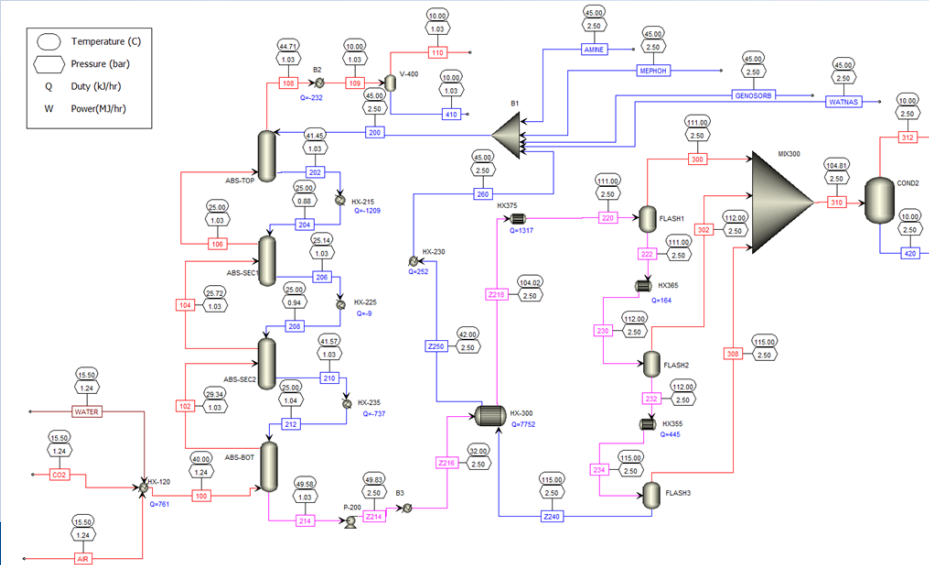
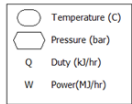
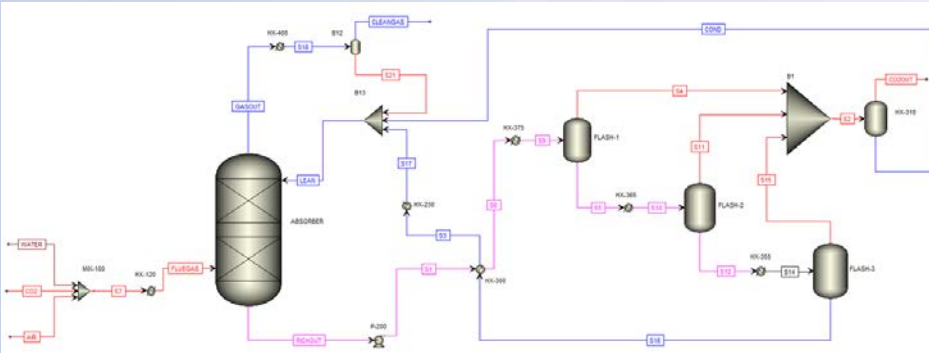


- In the absence of water kinetics are substantially slower than MEA
- With water, kinetics are approximately 2 times slower than MEA
- Ramifications
  - NAS requires higher absorber column to capture 90% CO<sub>2</sub> than 30wt% MEA
  - Process modelling of NAS showed a need for intercoolers to attain equilibrium
  - Use promoter to improve kinetics



- CPA-102 Calorimeter
- Stirred cell reactor
- Falling pressure drop method
- 260 mL reactor volume
- 22.6 cm<sup>2</sup> interfacial area
- T=298-353K
- P<sub>CO2</sub>= 4.48-6.29 kPa
- 100 mL solvent volume
- Kierzkowska-Pawlak et al., 2014, Int. J. Greenhouse Gas Control., 24, 106-114

# Process Modeling



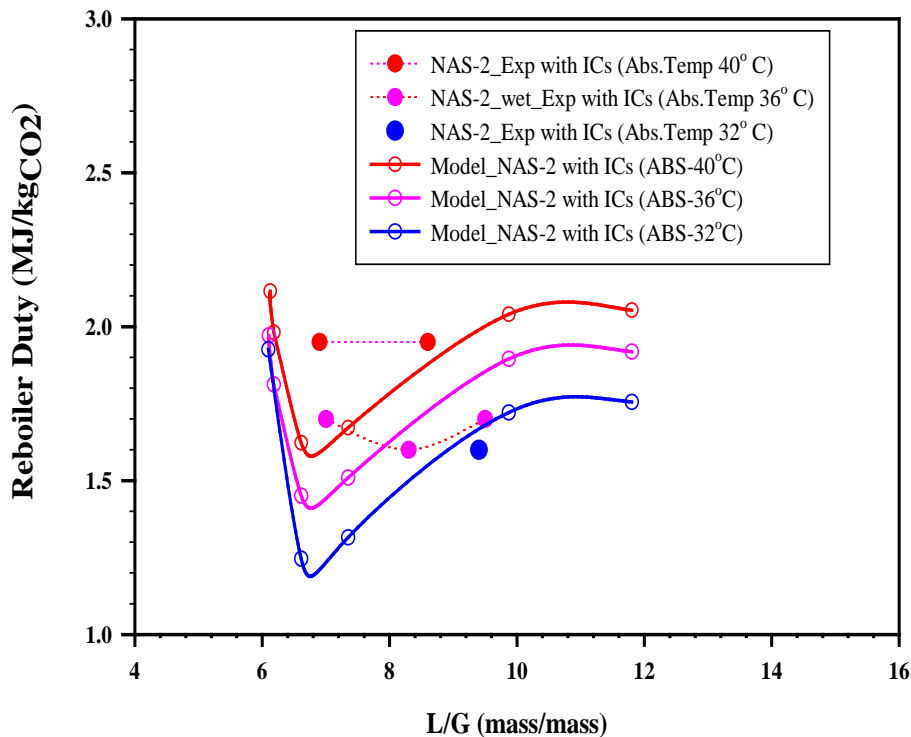
Developed rate-based process model  
Aspen ENRTL-SR

Thermodynamic and physical properties acquired experimentally:

- Henry's constant for CO<sub>2</sub>
- Liquid heat capacity
- Vapor pressures
- Reference state properties
- Heat of vaporization
- Dissociation constants
- VLE
- Density
- $\Delta h_{\text{abs}}$
- Viscosity
- Surface tension
- Thermal conductivity
- Dielectric constant
- Diffusivity of CO<sub>2</sub>

Used process model to direct bench-scale testing after initial runs

# Impact of Intercooler Temperatures on Reboiler Duty



- Impact of temperature on absorber bottom
- Modeled 40-32° C
- Lower temperature
- Lower L/G
- Lower reboiler duty
- Guided BsTU experiments at lower absorber temperatures
- Observed lower reboiler duties experimentally
- Will continue to investigate moving forward

## Conditions for Experimental Data

### NAS-2

- Absorber: 37-40° C
- Regenerator: 87-98° C
- Pressure: 2.5 bar
- Interstage Heater Regeneration

# Techno-Economic Analysis

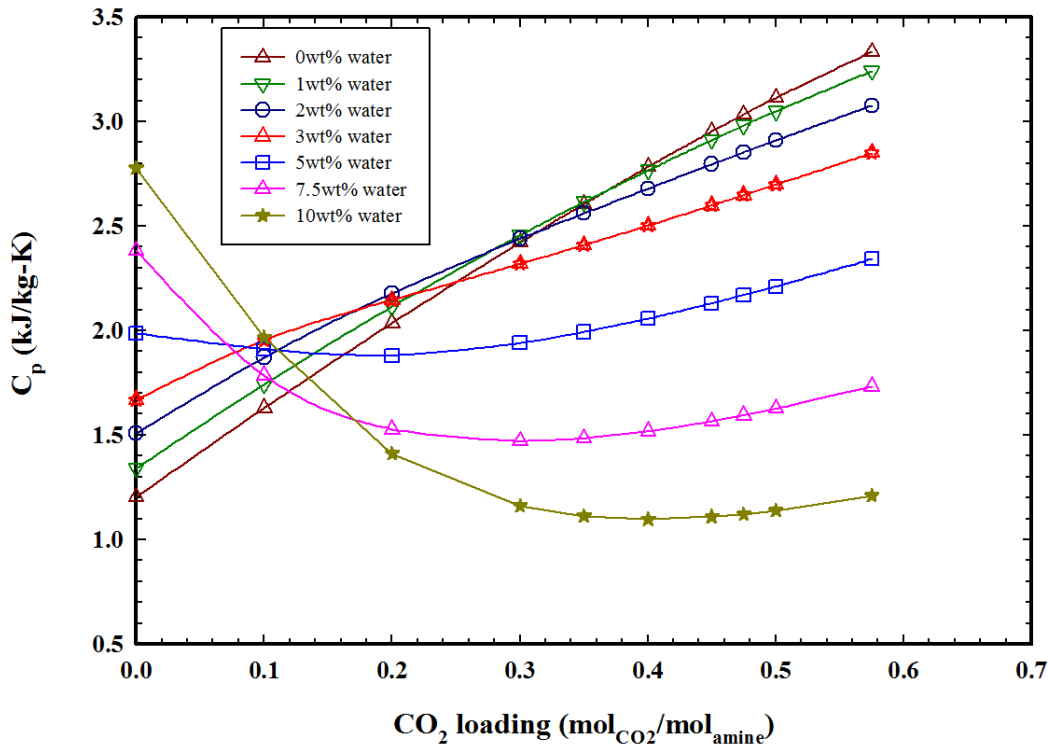
	Case 11_2011	Case 12_2011	NAS2-2.5 bar	NAS2-3.6 bar	NAS3-3.6 bar
	No Capture	w/ MEA			
<b>Power Performance</b>					
<i>Net Plant HHV Efficiency (%)</i>	39.3%	28.4%	31.10%	32.00%	32.90%
<b>Capital Investment (Total Installed Costs) 1000 \$</b>					
<i>Total Plant Cost (\$/kW)</i>	2,451	4,391	3,956	3,826	3,674
<b>Operating and Maintenance Costs</b>					
Annual (\$/y)	104,594,992	144,512,589	131,245,575	129,556,370	125,614,892
<b>COE Determination</b>					
<i>Total (\$/MWh)</i>	80.94	147.33	137.41	134.05	130.76
<i>ICOE (%)</i>	0%	82%	70%	66%	62%
<b>CO<sub>2</sub> Capture Summary</b>					
CO <sub>2</sub> Captured (tonne/MWh)		1.00	0.94	0.89	0.85
CO <sub>2</sub> Avoided (tonne/MWh)		0.69	0.71	0.70	0.70
CO <sub>2</sub> Capture Cost (\$/tonne)		66.7	59.8	59.9	58.7
CO <sub>2</sub> Capture Cost excl. TS&M (\$/tonne)		56.55	49.23	48.64	46.97
<i>CO<sub>2</sub> Avoided Cost (\$/tonne)</i>		96.0	80.0	74.1	73.0

## Summary of BP2 Testing

- With Linde, performed testing of NAS solvents in bench-scale test unit at 75-150 liter solvent scale using simulated flue gas
- Under dry conditions, measured reboiler heat duties as low as 2.4 GJ/tonCO<sub>2</sub> but did not realize duties as low as anticipated
- Under wet conditions, measured reboiler heat duties 1.6-1.9 GJ/tonCO<sub>2</sub> under conditions with regenerator operating at temperature less than 100° C
- Measured kinetics of CO<sub>2</sub> absorption and observed the rate constants of the wet solvent to be approximately 2 times slower than 30% aqueous MEA, with the kinetics of the dry solvent being substantially slower
- Developed rate-based ASPEN process model that matches well with experiment and used it to direct experiments
- Performed techno-economic analysis which shows potential of NAS process for lowering cost of CO<sub>2</sub> capture to ~\$47/ tonCO<sub>2</sub> (excluding TS&M costs)
- Completed long-term (five week) degradation testing at SINTEF on simulated flue gas showing that NAS is stable relative to aqueous MEA and is less corrosive

# Impact of water on NAS

**Predicted Liquid heat capacity of NAS-2 at 40°C  
at different water concentrations and CO<sub>2</sub> loadings**



## State-Point Data Table for NAS-1

	Units	Measured Performance	Projected Performance
<b>Pure Solvent</b>			
Molecular Weight	g mol <sup>-1</sup>	139.17 <sup>a</sup> 153.6 <sup>b</sup>	< 250
Normal Boiling Point	°C	243 to 288.45	<b>181 to 200</b>
Normal Freezing Point	°C	52.5 to -24	<b>52.5 to -24</b>
Vapor Pressure @ 15°C	Bar	0.00001 to 0.003 <sup>c</sup>	< 0.005 <sup>b</sup>
<b>Working Solution</b>			
Concentration	kg/kg	0.316 <sup>d</sup>	<b>0.4 to 0.6</b>
Specific Gravity (15°C)	kg/L	1.066 to 1.1 <sup>c</sup>	<b>0.9 to 1.2</b>
Specific Heat Capacity @ STP	kJ/kg K	1.28 to 1.48 <sup>d</sup>	<b>1.2 to 1.5</b>
Viscosity @ STP	cP	26.2 <sup>d</sup>	< 40
Surface Tension @ STP	dyn/cm	36.6 to 38.7 <sup>c</sup>	< 40
<b>Absorption</b>			
Pressure	bar CO <sub>2</sub>	0.133	<b>0.133</b>
Temperature	°C	35 to 45 (40)	<b>35 to 45</b>
Equilibrium Loading	g molCO <sub>2</sub> /kg	0.85 to 1.59 <sup>c</sup> (1.06)	<b>0.85 to 1.59</b>
Heat of Absorption	kJ/kg CO <sub>2</sub>	1,590 to 1,931 <sup>d</sup>	<b>1,590 to 1,931</b>
Solution Viscosity	cP	26.2	<b>2 to 30</b>
<b>Desorption</b>			
Pressure	bar CO <sub>2</sub>	2 to 7.8 (2.0)	<b>2 to 7.8</b>
Temperature	°C	90 to 120 (90)	<b>90 to 120</b>
Equilibrium Loading	g molCO <sub>2</sub> /kg	0.02 to 0.4 <sup>c</sup> (0.2)	<b>0.02 to 0.4</b>
Heat of Desorption	kJ/kg CO <sub>2</sub>	1,250 to 1,591 <sup>c</sup> (1,591)	<b>1,250 to 1,591</b>

<sup>a</sup> Nitrogenous Base Component

<sup>b</sup> NAS Formulation

<sup>c</sup> Individual components, range lowest to highest

<sup>d</sup> Ranges based on exp. measurements for most promising NASs

Italicized numbers used in preliminary technical and economic assessment.

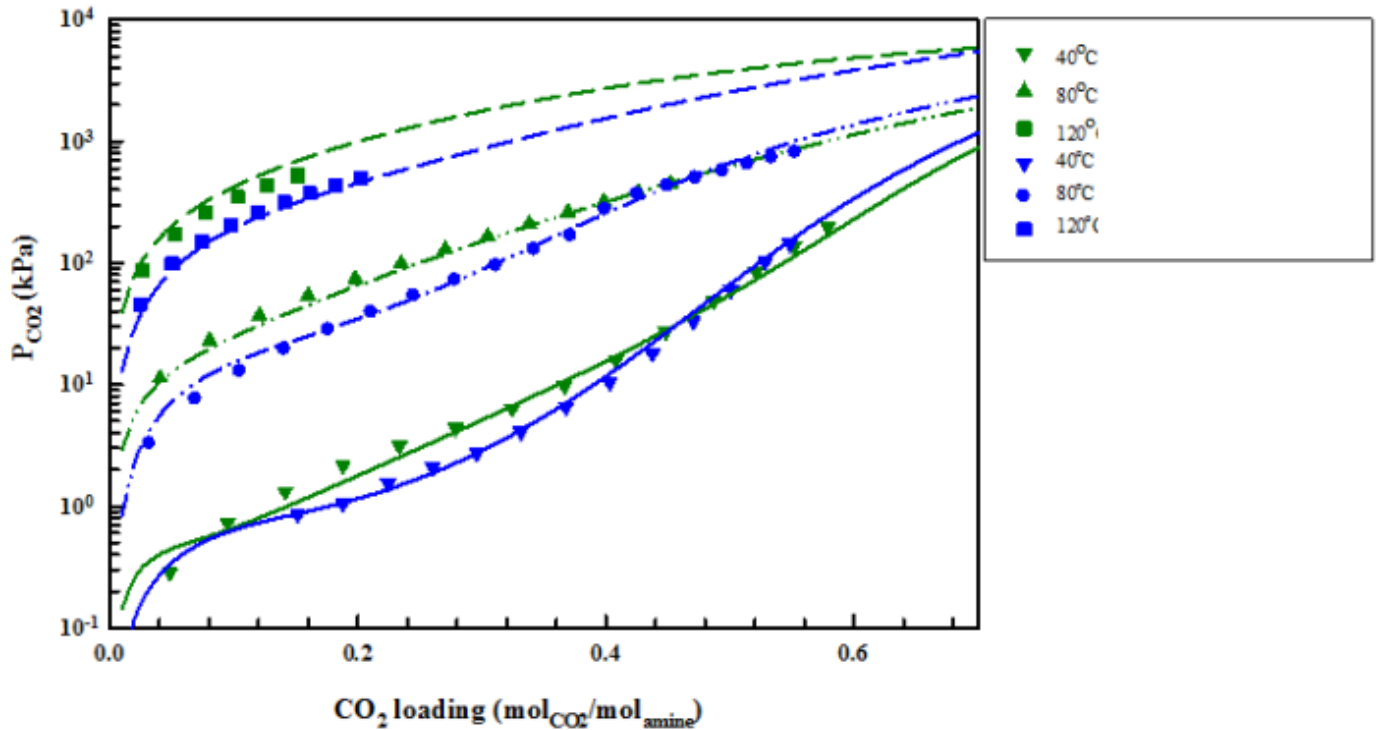


## Properties of NAS Solvent

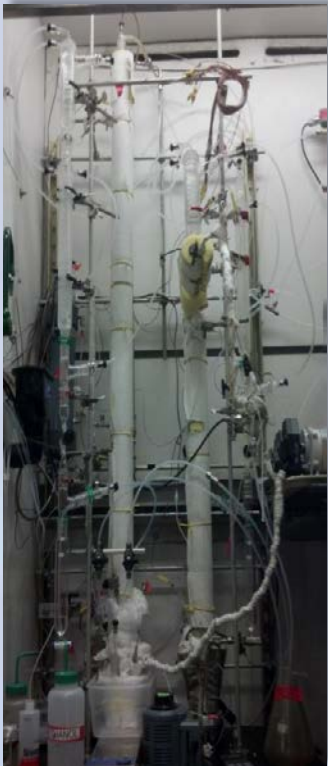
Criteria	Target	NAS-2
Vapor Pressure [kPa] @ 40°C	< 1	0.3 (Estimated)
Water Content [wt%]	<10	7.26
Viscosity [cP] CO <sub>2</sub> -rich at 40°C	< 40	< 30
Foaming Tendency	Low	Low
Cost [\$/kg]	< 50	comparable
Health Rating	≤ 3 (≤MEA)	2
Min. thermal regeneration energy* [kJt/kg CO <sub>2</sub> ]	<2,000	2,000

\*Notz et al. A short-cut method for assessing absorbents for post-combustion carbon dioxide capture. *Int. J. Greenhouse Gas Control* 2011, 5, 3 413-421

## Updated VLE Curves from ENRTL-SR



# Lab-Scale Gas Absorption System



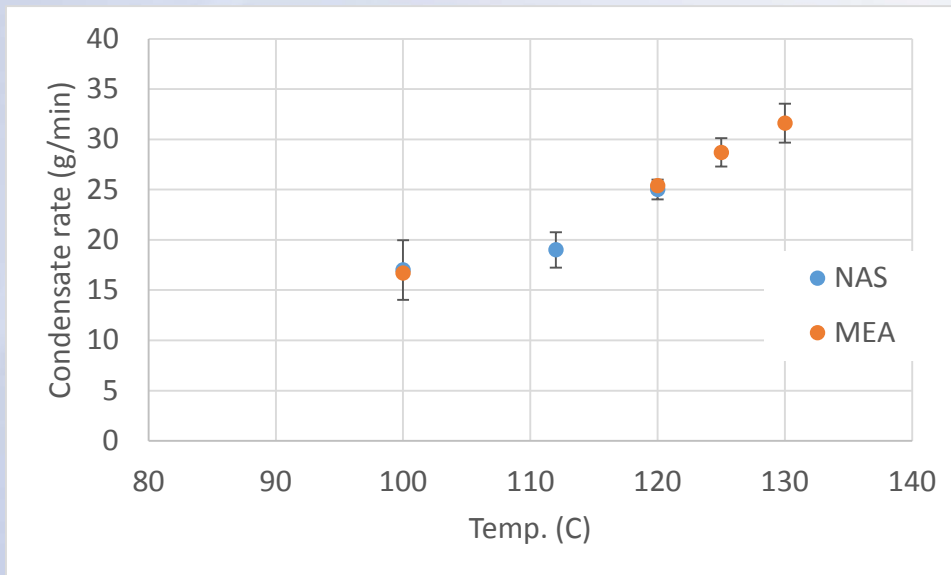
## Description

- Simple gas scrubbing system suitable for evaluation of aq. and non-aq. solvents
- 2-10 SLPM of sim. flue gas with relevant blends of  $\text{CO}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{O}_2$ ,  $\text{SO}_2$ ,  $\text{N}_2$
- Liquid flowrates of 10 to 130 mL/min
- Operates continuously; > 50 days (1,000h) commissioning with MEA-Water
- Total solvent volume: ~400 mL
- Off-line solvent compositional analysis
- On-line gas analysis

## Scope of Testing

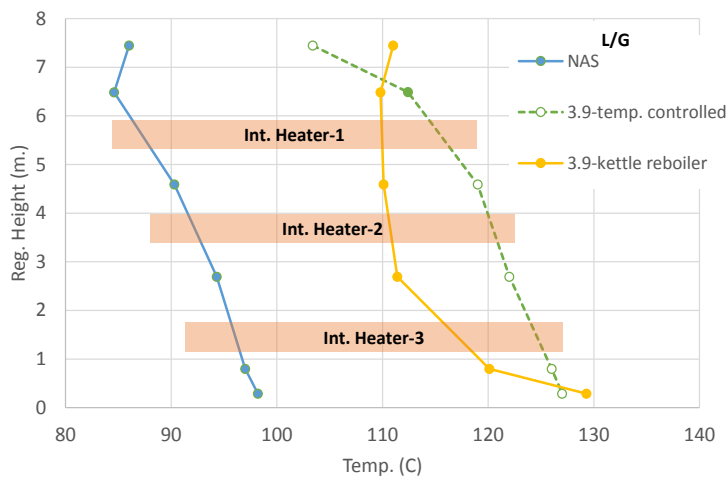
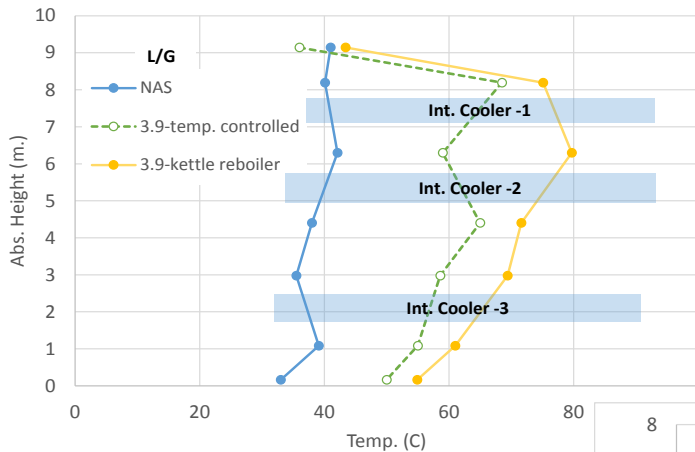
- Demonstrate stability of non-aq. solvents in a representative process arrangement using high-fidelity sim. FG
- Evaluate/demonstrate key process concepts specific to non-aqueous solvent process
- Compare performance of the NAS process and 30 wt% MEA- $\text{H}_2\text{O}$ 
  - Estimate regen. energy [ $\text{kJ}/\text{kg CO}_2$ ]
  - Support design of large, bench-scale unit

## Heat Loss Measurement BsTU



The heat loss determination using MEA/H<sub>2</sub>O solution was performed in similar manner as that of NAS where the regenerator was maintained at a uniform temperature of 100, 120, 125, and 130 °C while the lean MEA solution was circulating throughout the system. The heat loss measurement was evaluated for NAS at 100, 112, and 120 °C. The condensate collection during the heat loss determination at different temperature from both NAS and MEA solutions are provided in the figure.

# Temperature Profiles of the Absorber and Regenerator



## Kinetics Experiment

